

$-\text{RCO}_2\text{H}]^+$  (88) (calc. for  $\text{C}_{13}\text{H}_{18}\text{O}$ : 180.058), 152 [ $180 - \text{C}_2\text{H}_4]^+$  (26), 71 [ $\text{RCO}]^+$  (100); UV  $\lambda_{\text{max}}^{\text{Et}_2\text{O}}$  nm: 373, 347, 324, 304, 285, 272, 258, 239;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  3.74 and 3.71 (dd, H-1), 5.39 (ddd, H-2), 6.35 (dd, H-3), 5.83 (dd, H-4), 2.01 (s, H-13), OCOR: 2.61 (qq, H-2'), 1.20 (d, H-3', H-4'); ( $J$  [Hz]: 1,2=3.5; 1',2=2,3=6; 2,4=1; 2',3'=2',4'=7].

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#### REFERENCES

1. Cabrera, A. (1977) *The Biology and Chemistry of the Compositae* (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds), p. 1044. Academic Press, London.
2. Zdero, C., Bohlmann, F., King, R. M. and Robinson, H. (1986) *Phytochemistry* **25**, 2873.
3. Bohlmann, F., Banerjee, S., Jakupovic, J., King, R. M. and Robinson, H. (1985) *Phytochemistry* **24**, 1295.
4. Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1980) *Phytochemistry* **19**, 689.
5. Bohlmann, F., Ludwig, G. W., Jakupovic, J., King, R. M. and Robinson, H. (1984) *Lieb. Ann. Chem.* 228.
6. Bohlmann, F., Suwita, A., Jakupovic, J., King, R. M. and Robinson, H. (1981) *Phytochemistry* **20**, 1649.
7. Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1984) *Phytochemistry* **23**, 1979.

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## TINOSPORASIDE, AN 18-NORCLERODANE GLUCOSIDE FROM *TINOSPORA CORDIFOLIA*

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**Key Word Index**—*Tinospora cordifolia*; Menispermaceae; diterpene; 18-norclerodane glucoside; tinosporaside.

**Abstract**—The stem wood of *Tinospora cordifolia* has yielded a novel 18-norclerodane diterpene *O*-glucoside which has been assigned the trivial name tinosporaside. On the basis of extensive NMR studies this has been assigned the structure (relative stereochemistry) 1,17-dioxo-8 $\beta$ ,10 $\alpha$ ,12 $\alpha$ ,19 $\alpha$ ,20 $\beta$ -18-norclerod-2,13(16),14-trien-4 $\alpha$ - $\beta$ -D-glucopyranoside-12,17;15,16-dioxide.

#### INTRODUCTION

*Tinospora cordifolia* Miers has long been used in Ayurvedic medicine for the treatment of jaundice, rheumatism and urinary diseases [1]. Previous studies on this and other species of *Tinospora* [1–4] have revealed the presence of a range of clerodane-derived sesquiterpenes related to columbin (1) [5]. In this paper we report the isolation of an 18-norclerodane glucoside from the stem bark of *T. cordifolia*.

#### RESULTS AND DISCUSSION

Column chromatography of the ethyl acetate soluble part of an ethanolic extract of the stem bark gave, on repeated recrystallization from methanol, tinosporaside (2). High resolution EIMS indicated an empirical formula  $\text{C}_{25}\text{H}_{32}\text{O}_{10}$  with major fragments at  $m/z$  330 [ $\text{C}_{19}\text{H}_{22}\text{O}_5$ ] $^+$  and 314 [ $\text{C}_{19}\text{H}_{22}\text{O}_4$ ] $^+$  for loss of a hexose sugar moiety (as an *O*-glycoside) with, respectively, retention and loss of the ether oxygen. The IR spectrum revealed the presence of two carbonyl bands at  $1708\text{ cm}^{-1}$  ( $\gamma$ -lactone) and  $1670\text{ cm}^{-1}$  ( $\alpha,\beta$ -unsaturated carbonyl).

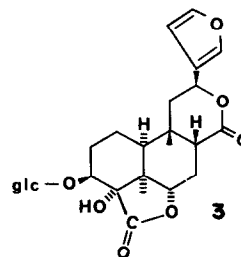
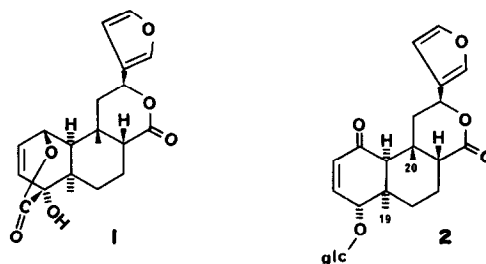


Table 1.  $^1\text{H}$  and  $^{13}\text{C}$  NMR chemical shift data for 2-Ac<sub>4</sub> and  $^{13}\text{C}$  data for 1 [6] and 3-Ac<sub>5</sub> [2]

C	$^1\text{H}$ spectrum ( <i>J</i> values: Hz)	$^{13}\text{C}$ spectrum		
	2-Ac <sub>4</sub>	1	2-Ac <sub>4</sub>	3-Ac <sub>5</sub>
1	—	74.1	201.8	16.9
2	5.99 <i>d</i> (10.3)	136.7	129.4	25.4
3	6.70 <i>ddd</i> (10.3, 4.9, 1.3)	128.9	141.8	78.5
4	4.22 <i>d</i> (4.9)	80.4	72.0	87.9
5	—	37.1	43.3	46.9
6 <sub>eq</sub>	2.25 <i>ddd</i> (14.3, 4.0, 4.0)	25.6	29.2	70.6
6 <sub>ax</sub>	1.02 <i>ddd</i> (14.3, 14.0, 3.8)			
7 <sub>eq</sub>	2.19 <i>m</i>	17.3	18.9	28.4
7 <sub>ax</sub>	1.63 <i>dddd</i> (14.3, 14.1, 8.3, 1.5)			
8	2.30 <i>br s</i>	47.6	48.7	46.3
9	—	35.2	35.5	36.3
10	2.17 <i>s</i>	44.4	48.5	46.1
11 <sub>eq</sub>	2.25 <i>dd</i> (15.0, 3.6)	41.9	40.6	44.0
11 <sub>ax</sub>	1.82 <i>dd</i> (15.0, 12.2)			
12	5.47 <i>dd</i> (12.2, 3.6)	70.6	69.8	70.6
13	—	124.7	124.6	124.4
14	6.41 <i>dd</i> (1.4, 0.9)	108.3	108.0	108.3
15	7.43 <i>t</i> (1.4)	143.8	144.7	139.8
16	7.52 <i>t</i> (0.9)	139.5	139.5	143.8
17	—	173.1	171.0*	172.1
18	—	175.3	—	171.4
19	1.18 <i>s</i>	28.2	31.1	33.4
20	0.80 <i>s</i>	24.3	25.7	18.5
1'	4.65 <i>d</i> (7.7)	—	100.3	101.9
2'	4.94 <i>dd</i> (9.5, 7.8)	—	71.4	71.8
3'	5.18 <i>t</i> (9.4)	—	72.2	74.5†
4'	5.04 <i>t</i> (9.6)	—	68.0	68.6
5'	3.65 <i>ddd</i> (10.0, 4.6, 2.9)	—	72.0	72.3†
6'	4.14 <i>dd</i> (12.4, 4.8)	—	61.6	61.8
	4.11 <i>dd</i> (12.4, 2.8)			
Ac	2.01/1.99/1.97/1.97	—	170.2 */20.5 170.0*/20.4 169.9*/20.3 168.7*/20.3	170.4†/21.7 169.9†/20.9 169.7†/20.6 169.4†/20.5 168.7†

All spectra run in CDCl<sub>3</sub>.  $^1\text{H}$  NMR at 360 MHz for 2;  $^{13}\text{C}$  NMR at 62.5 MHz for 1, 90.56 MHz for 2-Ac<sub>4</sub> and 22.5 MHz for 3-Ac<sub>5</sub>.

\*†Assignments are interchangeable.

Peracetylation yielded a tetra-acetate (2-Ac<sub>4</sub>) the  $^1\text{H}$  NMR spectrum of which allowed resolution of all 40 protons (Table 1). Coupling constants for the methine protons H-1' to H-5' of the hexose showed an all *trans*-axial relationship and together with the methylene (H-6') resonances confirmed the identity of the sugar as  $\beta$ -glucose. Hydrolysis of 2 gave a hexose ( $[\alpha]_{\text{D}} + 49^\circ$ ), thus establishing the occurrence of  $\beta$ -D-glucose.

Olefinic resonances in the  $^1\text{H}$  NMR spectrum at  $\delta$  6.41, 7.43 and 7.52 were typical for a 3-substituted furan system and an isolated ABX system at  $\delta$  5.47, 1.82 and 2.25 could be assigned to H-12 and H-11 protons with H-12 axial. Both patterns are closely comparable to the spectrum of compound 1 [6].

Two further olefinic protons resonated  $\delta$  5.99 and 6.70, typical of the  $\alpha$ - and  $\beta$ -protons of an  $\alpha,\beta$ -unsaturated ketone. The  $\beta$ -proton showed an additional coupling to a proton resonating at  $\delta$  4.22, which can be assigned to the carbinolic position of glycosidic attachment. These data require the system  $-\text{CO}-\text{CH}=\text{CH}-\text{CH}(\text{O-glucose})-\text{CR}^3$

which can be attributed to the A-ring of a columbin-like clerodane diterpene in which the 18-methyl group has been lost. A sharp singlet at  $\delta$  2.17 could be assigned to H-10. A second, broadened singlet at  $\delta$  2.30 must be attributed to H-8. This is unusual; in compound 1 H-8 appears as a double-doublet with a large coupling to one of the H-7 protons [6]. In 2-Ac<sub>4</sub> the axial H-6 and H-7 protons are clearly visible as shielded resonances, that for H-7<sub>ax</sub> showing two large ( $J_{\text{gem}}$  and  $J_{6\text{ax}}$ ) and two small ( $J_{6\text{eq}}$  and  $J_8$ ) couplings. Thus the stereochemistry of the B-ring of 2 appears to be a normal chair in which H-8 is equatorial. This contrast with 1 where it is considered to occur in the boat conformation [6].

The  $^{13}\text{C}$  NMR spectrum of 2-Ac<sub>4</sub> was obtained and correlated with the proton spectrum. A comparison with that obtained for compound 1 (Table 1) showed the anticipated differences related to changes in A-ring substituents. One striking feature was the deshielded nature of C-19 ( $\delta$  31.1) which is similar to that recorded for boreptoside-A (3) (Table 1) isolated from *Tinospora tuber-*

Table 2. NOE enhancements for 2-Ac<sub>4</sub>

Irradiation	Enhancement (%)
H-4	H-3 (3), H-1' (1), H-6 <sub>eq</sub> (1)
H-7 <sub>ax</sub>	H-7 <sub>eq</sub> (3.5), H-6 <sub>eq</sub> (1)
H-11 <sub>ax</sub>	H-11 <sub>eq</sub> (2.5), H-8 (2.5)
H-12	H-10 (5), H-16 (3)
Me-19	H-10 (3.5)
Me-20	H-2 (2), H-4 (4), H-7 <sub>ax</sub> (8), H-8 (3), H-11 <sub>ax</sub> (7)

*culata* [2]. The C-20 methyl resonance is also somewhat deshielded in comparison with many similar clerodanes. This appears to be attributable to its relative proximity to the C-2/C-3 double-bond.

The stereochemistry of compound **2** was further explored through a series of NOE experiments (Table 2). The key observations were as follows: (i) a strong interaction between H-12 and H-10; (ii) H-11<sub>ax</sub> (on the opposite face of the molecule to H-12) interacts with H-8. Together these findings indicate H-8 and H-10 are on opposite faces and that H-8 and H-11<sub>ax</sub> are 1,3-diaxial. (iii) Irradiation of Me-20 caused strong enhancement of H-11<sub>ax</sub>, H-8, H-7<sub>ax</sub> and H-4. This requires the anticipated *cis* B/C ring junction and places H-4 on the same face as Me-20. (iv) Irradiation of Me-19 causes a strong enhancement of H-10 confirming that the A/B ring junction is also *cis*.

On this basis the glycoside was assigned structure **2** or its enantiomer and it has been given the trivial name tinosporaside. The dextrorotatory nature of **2** suggests the stereochemistry is as depicted. Compound **2** is obviously related to **3** and shows comparable stereochemistry for both A/B and B/C ring junctions. Surprisingly, a similar diterpene reported from *T. cordifolia* [1] has been assigned a *trans* B/C ring junction although the reasoning for this is not clear and the <sup>13</sup>C NMR resonance for C-11 (δ 41.9) is similar to that found in compounds **1–3**. By contrast in floribundic acid, where the B/C ring fusion is *trans*, C-11 resonates at (δ 45.8).

#### EXPERIMENTAL

**Plant material.** The stem bark of *T. cordifolia* was collected from the outskirts of Hyderabad. A voucher specimen (No. 63)

has been deposited at the Herbarium of the Department of Botany, Osmania University, Hyderabad, India.

**Extraction and isolation of compound 2.** Air-dried stem bark (5 kg) was defatted with petrol (bp 60–80°) and then extracted to exhaustion (Soxhlet) with EtOH. The extract was concd and EtOAc added. After cooling a brown solid was formed which was collected and refluxed several times with C<sub>6</sub>H<sub>6</sub>. The C<sub>6</sub>H<sub>6</sub> insoluble material was then subjected to CC eluting with 5% MeOH in CHCl<sub>3</sub> to give **2** which was recrystallized from aq. MeOH as needles (2 g), mp 228–229°, [α]<sub>D</sub> + 65° (MeOH; c 1.00). UV λ<sub>max</sub> nm: 218. IR ν<sub>max</sub> cm<sup>-1</sup>: 3480–3400, 1708, 1670, 1075, 1022. EIMS *m/z* (rel. int.): 492.2017 [M]<sup>+</sup> (41) (C<sub>25</sub>H<sub>32</sub>O<sub>10</sub> requires 492.1892); 330 [C<sub>19</sub>H<sub>22</sub>O<sub>5</sub>]<sup>+</sup> (58), 314 [C<sub>19</sub>H<sub>22</sub>O<sub>4</sub>]<sup>+</sup> (32), 312 (61), 205 (72), 204 (52), 121 [C<sub>8</sub>H<sub>9</sub>O]<sup>+</sup> (100).

**Acetylation of 2** (200 mg) was achieved by refluxing with Ac<sub>2</sub>O (2 ml) in pyridine (2 ml) for 7 hr. Normal work-up and recrystallization from MeOH gave 2-Ac<sub>4</sub>, mp 218–219°. IR ν<sub>max</sub> cm<sup>-1</sup>: 1750, 1730, 1680, 1230, 1210, 1040. <sup>1</sup>H and <sup>13</sup>C NMR, see Table 1.

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#### REFERENCES

1. Hanuman, J. B., Bhatt, R. K. and Sabata, B. K. (1986) *Phytochemistry*, **25**, 1677.
2. Fukuda, N., Yonemitsu, M., Kimura, T., Hachiyama, S., Miyahara, K. and Kawasaki, T. (1985) *Chem. Pharm. Bull.* **33**, 4438.
3. Ahmad, M., Kazi, A. B., Karim, R., Khaleque, A. and Miah, M. A. W. (1978) *J. Bangladesh Acad. Sci.* **2**, 25.
4. Bhatt, R. K., Hanuman, J. B. and Sabata, B. K. (1988) *Phytochemistry* **27**, 1212.
5. Overton, K. H., Weir, N.G. and Wylie, A. (1966) *J. Chem. Soc. C*, 1482.
6. Waterman, P. G., Reza-ul-Zalil, Hasan, C. M. and Jabbar, A. (1985) *Planta Med.*, 181.